

SYNTHESIS OF NEW 1,8-DIHYDROXY-9H-XANTHEN-9-ONES

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INTRODUCTION

Xanthenes have a rather restricted occurrence among higher plants, being found almost exclusively in *Guttiferae* and *Gentianaceae*. Xanthenes are sometimes found as the polyhydroxylated compounds but more often with a varying degree of substitution at the phenyl rings. However, many xanthenes bearing hydroxyl substituents exhibit valuable biological activity.¹

Xanthenes are well known for their interesting phytochemical properties, which make them attractive to the pharmaceutical industry. Isolated or synthetic ones have been reported to exhibit several important biological activities, such as anti-tumor,² anti-inflammatory,³ antioxidant⁴ and anticoagulant/antiplatelet agents.⁵



Guttiferae



Gentianaceae

Owing to our continuing interest in the synthesis of novel oxygen heterocycles we have recently developed unique synthetic procedure towards the synthesis of new (*E*)-4-benzylidene-8-hydroxyl-3-phenyl-3,4-dihydro-1*H*-xanthene-1,9(2*H*)-dione **2** derivatives.⁶

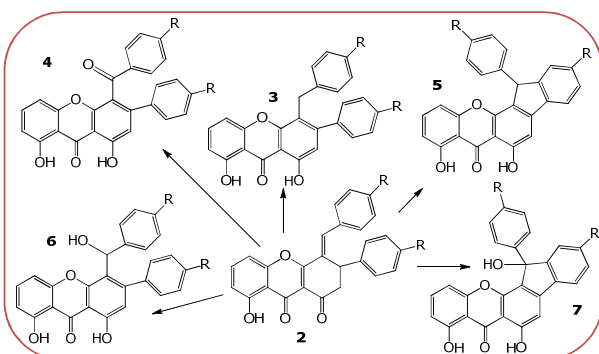
We set up a program aiming the transformation of the xanthenedione derivatives **2** into new 1,8-dihydroxy-9*H*-xanthen-9-one **3** derivatives.

RESULTS and DISCUSSION

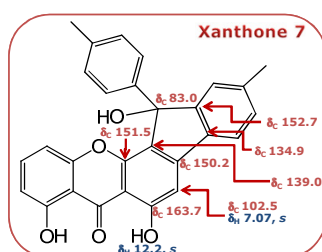
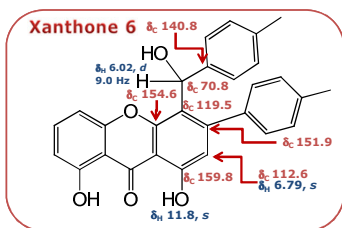
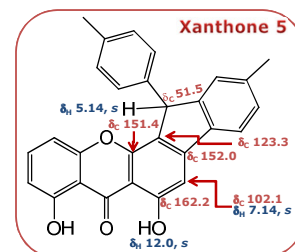
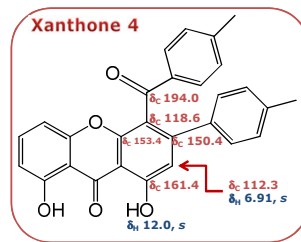
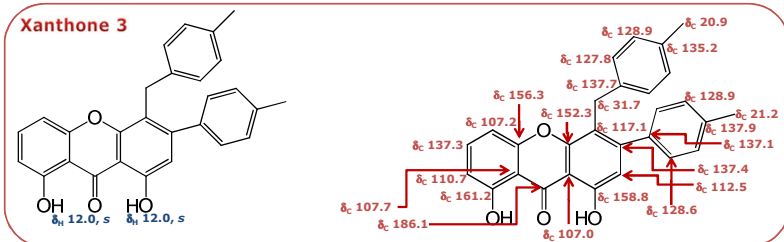
The use of acid (H₂SO₄, *p*-TSA) or base (DBU, LiHMDS) catalysis or even in oxidant conditions (DMSO/I₂) at room temperature, with different amounts of catalyst and trying different reaction times, didn't perform the desired aromatization. Decomposition products and mainly started reagent **2** were obtained.

Next several conditions using microwave irradiation and classic heating conditions were tested (Table). The desired product was obtained in low yields and the best results were obtained when DBU was used (Table, entry 5).

Instead several other new derivatives were isolated (structures **4-7**) and their structures confirmed mainly by NMR experiments.



Entry	Catalyst	Experimental condition	Obtained compound (%)
1	2 eq DDQ	MW (170°, 30 min) TCB acid	4 (28%)
2	1.5 eq DDQ	MW (170°, 30 min) dry TCB acid, molecular sieves	5 (52%)
3	1.3 eq DDQ	MW (100°, 30 min) dry TCB acid, molecular sieves	4 (15%) 6 (30%)
4	2 eq ChA	MW (170°, 30 min) dry TCB acid, molecular sieves	4 (10%) 7 (34%)
5	1.2 eq DBU	MW (100°, 10 min) DMSO	3 (18%)
6	LiHMDS	Classical heating (80°)	3 (< 5%)



CONCLUSIONS

Xanthenedione **2** revealed to be an interesting key compound to obtain new xanthenone derivatives **3-5** that can possess special biological and medicinal properties;

Currently we are trying to improve the yields in order to obtained these new derivatives in amounts that allowed biological assessments.

REFERENCES

- O. Demirkiran, *Top Heterocycl. Chem.* **2007**, 9, 139-178.
- M. Pedro, F. Cerqueira, M. E. Sousa, M. S. J. Nascimento, M. Pinto, *Bioorg. Med. Chem.* **2002**, 10, 3725-3730.
- H. H. Park, Y.-D. Park, J.-M. Han, K.-R. Im, B. W. Lee, I. Y. Jeong, T.-S. Jeong, W. S. Lee, *Bioorg. Med. Chem. Lett.* **2006**, 16, 5580.
- P. Suvarnakuta, C. Chaweerungrat, S. Devahastin, *Food Chemistry* **2011**, 125, 240.
- M. Correia-da-Silva, E. Sousa, B. Duarte, F. Marques, F. Carvalho, L. M. Cunha-Ribeiro, M. M. M. Pinto, *J. Med. Chem.* **2011**, 54, 5373-5384.
- D. C. G. A. Pinto, A. M. L. Seca, S. B. Leal, A. M. S. Silva, J. A. S. Cavaleiro, *Synlett*, **2011**, 2005-2008.

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